#### Balancing Chemical Equations Worksheet

3. 
$$S_8 + 12 O_2 \rightarrow 8 SO_3$$

4. 
$$2 N_2 + 0_2 \rightarrow 2 N_20$$

5. 
$$2 \text{ HgO} \rightarrow 2 \text{ Hg} + 0_2$$

6. 
$$\underline{-6}$$
 CO<sub>2</sub> +  $\underline{-6}$  H<sub>2</sub>O  $\Rightarrow$   $\underline{-}$  C<sub>6</sub>H<sub>12</sub>O<sub>6</sub> +  $\underline{-6}$  O<sub>2</sub>

7. 
$$Zn + Z HCI \rightarrow ZnCl_2 + H_2$$

9. 
$$_{2}$$
 Na +  $_{2}$  H<sub>2</sub>O  $\rightarrow$   $_{2}$  NaOH +  $_{1}$  H<sub>2</sub>

10. 
$$2 H_3PO_4 \rightarrow H_4P_2O_7 + H_2O_7$$

12. \_\_\_\_ 
$$CO_2$$
 + \_\_\_\_  $NH_3 \rightarrow$  \_\_\_\_  $OC(NH_2)_2$  + \_\_\_\_  $H_2O$ 

13. 
$$4 \text{ Si}_2\text{H}_3 + 70 \text{ O}_2 \rightarrow 8 \text{ SiO}_2 + 6 \text{ H}_2\text{O}_3$$

14. 
$$2 \text{ Al}(OH)_3 + 3 \text{ H}_2SO_4 \rightarrow Al_2(SO_4)_3 + 6 \text{ H}_2O$$

16. \_\_\_\_ Fe<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub> + \_\_\_6\_ KOH 
$$\rightarrow$$
 \_\_\_3\_ K<sub>2</sub>SO<sub>4</sub> + \_\_\_2\_ Fe(OH)<sub>3</sub>

17. 
$$2 C_7H_6O_2 + 15 O_2 \rightarrow 14 CO_2 + 5 H_2O$$

19. \_\_\_\_\_ FeS<sub>2</sub> + \_\_\_\_\_ O<sub>2</sub> 
$$\rightarrow$$
 \_\_\_\_\_ Fe<sub>2</sub>O<sub>3</sub> + \_\_\_\_\_ SO<sub>2</sub>

21. \_\_\_\_ Fe<sub>2</sub>O<sub>3</sub> + \_\_3\_ H<sub>2</sub> 
$$\rightarrow$$
 \_\_2\_ Fe + \_\_3\_ H<sub>2</sub>O

22. \_\_\_\_ Na<sub>2</sub>CO<sub>3</sub> + \_\_2\_ HCl 
$$\rightarrow$$
 \_2\_ NaCl + \_\_\_\_ H<sub>2</sub>O + \_\_\_\_ CO<sub>2</sub>

24. \_\_\_\_ 
$$C_7H_{16} + //_ O_2 \rightarrow //_ CO_2 + //_ H_2O$$

25. 
$$P_4 + 5 O_2 \rightarrow 2 P_2O_5$$

# **Balancing Chemical Equation Answers**

**University of the State of New York** 

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