

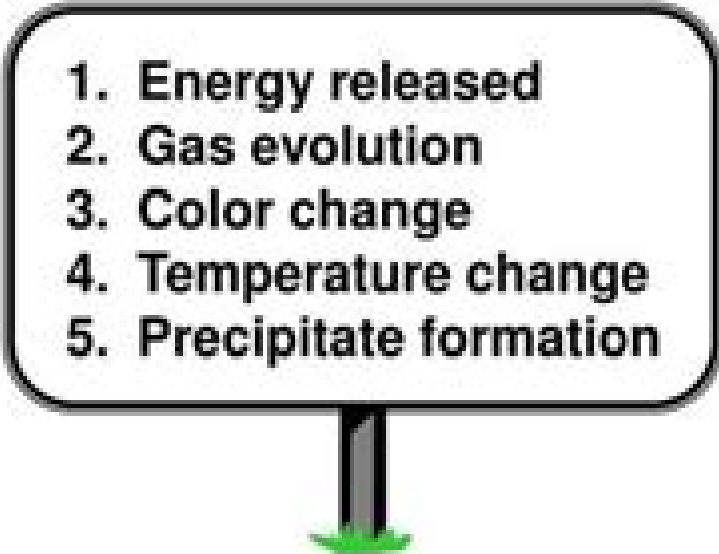
The solubility rules allow us to determine if a double-replacement reaction will actually occur.

Consider:



Did a reaction actually occur?
(Hint: Think about the 5 indicators of chemical change)

No, there is no reaction because all of the ions remain in solution.

- 
1. Energy released
 2. Gas evolution
 3. Color change
 4. Temperature change
 5. Precipitate formation

How about:



Yes, because a precipitate was formed.

Double Replacement And Solubility Pogil Answers

CH Cherryholmes



Double Replacement And Solubility Pogil Answers:

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