



Reinforcement

Structure of the Atom

Answer the questions that following statements are giving correctly or not.

Q. 1. How is the electronic configuration of an element determined?

Q. 2. What are groups represented?

Q. 3. How electronic property is connected the number of valence electrons, why not?

Q. 4. Which among factors of periodic structure contribute to the extent of size of atom?

Q. 5. For the elements in same periods, why is atomic radius?

Q. 6. For the two, the periodic trends changed, why then?

Q. 7. For the elements in same period, why is the IE (ionization) trend observed?

Answer the following and write in correct manner. Particular is required to write the right thing correct answers in the blank of the left.

Question 12

- _____ a. electron
- _____ b. neutron
- _____ c. neutron
- _____ d. proton
- _____ e. alpha

Question 13

- a. positively charged particle
- b. negatively charged particle
- c. non-particle
- d. smaller particles than matter up particle configuration
- e. positively charged particle of an atom

Chapter 10 Reinforcement Structure Of The Atom

Answer Key

B Lingard



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