

9 • Bonding & Molecular Structure

LEWIS STRUCTURE PRACTICE - 1

Ca^{2+} $[Ca]^{2+}$	NO_3^- $\leftrightarrow \left[\begin{array}{c} \text{O}^- - \text{N} = \text{O} \\ \\ \text{O}^- \end{array} \right]^- \leftrightarrow$	PCl_5
C_2H_2 $\text{H} : \text{C} :: \text{C} : \text{H}$	S^{2-} $[\text{:}\ddot{\text{S}}\text{:}]^{2-}$	NH_3 $\text{H} : \ddot{\text{N}} : \text{H}$ $ $ H
SO_4^{2-} $\left[\begin{array}{c} \text{O} \\ \text{:}\ddot{\text{O}}\text{:} \\ \text{:}\ddot{\text{S}}\text{:} \\ \text{:}\ddot{\text{O}}\text{:} \\ \text{O} \end{array} \right]^{2-}$	SO_2 $\text{O} = \ddot{\text{S}} = \text{O} \leftrightarrow \ddot{\text{O}} = \ddot{\text{S}} - \text{O}^-$	CaH_2 $\text{H} : \text{Ca} : \text{H}$



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LEWIS STRUCTURE PRACTICE - 2

CH_4 $\text{H} - \text{C} - \text{H}$ $ $ H	HCN $\text{H} : \text{C} :: \text{N} :$	K^+ $[\text{K}]^+$
PO_4^{3-} $\left[\begin{array}{c} \text{O} \\ \text{:}\ddot{\text{O}}\text{:} \\ \text{:}\ddot{\text{P}}\text{:} \\ \text{:}\ddot{\text{O}}\text{:} \\ \text{O} \end{array} \right]^{3-}$	O_3 $\text{O} = \text{O} - \text{O} \leftrightarrow \ddot{\text{O}} = \ddot{\text{O}} - \ddot{\text{O}}$	SF_4
Br^- $[\text{:}\ddot{\text{Br}}\text{:}]^-$	AlH_3 $\text{H} : \text{Al} : \text{H}$ $ $ H	NH_4^+ $\left[\begin{array}{c} \text{H} \\ \text{H} : \text{N} : \text{H} \\ \\ \text{H} \end{array} \right]^+$

Bonding And Molecular Structure Worksheet Answers

H Kauffman



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