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### WORKSHEET: Chemical Bonding – Ionic & Covalent!

#### **REMEMBER...**

**Ionic Bond**

between a Metal and Non-Metal

(M + NM)

**Covalent Bond**

between a Non-Metal and Non-Metal

(NM + NM)

**PART 1:** Determine if the elements in the following compounds are metals or non-metals. Describe the type of bonding that occurs in the compound.

Compound	Element 1 (metal or non-metal?)	Element 2 (metal or non-metal?)	Bond Type
NO <sub>2</sub>	N = non-metal	O = non-metal	covalent
NaCl	Na = Metal (M)	Cl = Nonmetal (NM)	I
SO <sub>2</sub>	S = NM	O = NM	C
PI <sub>3</sub>	P = NM	I = NM	C
MgBr <sub>2</sub>	Mg = M	Br = NM	I
CaO	Ca = M	O = NM	I
H <sub>2</sub> O	H = NM	O = NM	C
K <sub>2</sub> O	K = M	O = NM	I
AlF <sub>3</sub>	Al = M	F = NM	I
O <sub>2</sub>	O = NM	O = NM	C
CuCl <sub>2</sub>	Cu = M	Cl = NM	I
NO <sub>2</sub>	N = NM	O = NM	C
CO <sub>2</sub>	C = NM	O = NM	C
HF	H = NM	F = NM	C
Rb <sub>2</sub> S	Rb = M	S = NM	I
NBr <sub>3</sub>	N = NM	Br = NM	C
Fe <sub>2</sub> O <sub>3</sub>	Fe = M	O = NM	I
CCl <sub>4</sub>	C = NM	Cl = NM	C

# Bonding Ionic Or Convalent Packet Answers

**Kenneth W. Raymond**



## **Bonding Ionic Or Covalent Packet Answers:**

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exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students

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## **Table of Contents Bonding Ionic Or Convalent Packet Answers**

1. Understanding the eBook Bonding Ionic Or Convalent Packet Answers
  - The Rise of Digital Reading Bonding Ionic Or Convalent Packet Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Bonding Ionic Or Convalent Packet Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Bonding Ionic Or Convalent Packet Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Bonding Ionic Or Convalent Packet Answers
  - Personalized Recommendations
  - Bonding Ionic Or Convalent Packet Answers User Reviews and Ratings
  - Bonding Ionic Or Convalent Packet Answers and Bestseller Lists
5. Accessing Bonding Ionic Or Convalent Packet Answers Free and Paid eBooks

- Bonding Ionic Or Covalent Packet Answers Public Domain eBooks
  - Bonding Ionic Or Covalent Packet Answers eBook Subscription Services
  - Bonding Ionic Or Covalent Packet Answers Budget-Friendly Options
6. Navigating Bonding Ionic Or Covalent Packet Answers eBook Formats
- ePub, PDF, MOBI, and More
  - Bonding Ionic Or Covalent Packet Answers Compatibility with Devices
  - Bonding Ionic Or Covalent Packet Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Bonding Ionic Or Covalent Packet Answers
  - Highlighting and Note-Taking Bonding Ionic Or Covalent Packet Answers
  - Interactive Elements Bonding Ionic Or Covalent Packet Answers
8. Staying Engaged with Bonding Ionic Or Covalent Packet Answers
- Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Bonding Ionic Or Covalent Packet Answers
9. Balancing eBooks and Physical Books Bonding Ionic Or Covalent Packet Answers
- Benefits of a Digital Library
  - Creating a Diverse Reading Collection Bonding Ionic Or Covalent Packet Answers
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Bonding Ionic Or Covalent Packet Answers
- Setting Reading Goals Bonding Ionic Or Covalent Packet Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Bonding Ionic Or Covalent Packet Answers
- Fact-Checking eBook Content of Bonding Ionic Or Covalent Packet Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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