

Chapter

6

CHEMICAL BONDING

MULTIPLE CHOICE QUESTIONS

- Which of the following molecules has a coordinate covalent bond?
(a) HCl (b) NaCl (c) NH_4Cl (d) AlCl_3 [LHR 05]
- The angle formed in sp hybridization is:
(a) 120° (b) 180° (c) 109.5° (d) 107.5° [GRW 06]
- Ionic compounds are mostly obtained by the combination of group:
(a) 3 and 5 (b) 2 and 5 (c) 4 and 8 (d) 1 and 7 [GRW 07]
- Dipole moment of CO_2 is:
(a) 1.84D (b) Zero D (c) 0.95 D (d) 2.2 D [GRW 07]
- In sp^2 hybridization, the orbitals are oriented at an angle of:
(a) 109.5° (b) 120° (c) 180° (d) 0° [LHR 07]
- Which of the following species has unpaired electron in anti bonding molecular orbitals?
(a) H_2 (b) He_2 (c) O_2^{+2} (d) N_2^{+2} [MTN 08, 10, SWP, LHR 08]
- A molecular orbital can contain maximum electrons equal to:
(a) One (b) Two (c) Three (d) Four [LHR 08]
- Carbon dioxide and methane have dipole moment value:
(a) Zero and 1.85 D (b) 1.70 D and 1.80 D (c) Both have zero (d) None of these [GRW 08]
- Nobel gases have maximum stability and least reactive because:
(a) They are very safe (b) Their valence shells are complete
(c) They are gases (d) They are present in zero group [GRW 07]
- Which species has unpaired electron in anti-bonding molecular orbits?
(a) O_2^{+2} (b) N_2^{+2} (c) B_2 (d) F_2 [FSD 11, GRW 09, 12]
- Octet rule is not followed in the formation of:
(a) NF_3 (b) CF_4 (c) CCl_4 (d) PCl_5 [LHR 09]
- The most stable elements are:
(a) Halogens (b) Lithium family (c) Noble gases (d) None of these [GRW 09, LHR 14]
- The hybridization of carbon in C_2H_4 is:
(a) sp (b) sp^2 (c) sp^3 (d) not hybridized [LHR 09]
- Positive ions are called:
(a) Cations (b) Anions (c) Molecules (d) Hydrated ions [LHR 11]
- Total number of bonds in C_2H_4 molecule is:
(a) Six (b) Four (c) Five (d) Eight [LHR 10, 11]
- The S.I unit of dipole moment is:
(a) Joule (b) Debye (c) Coulomb meter (d) Nm^{-2} [GRW 11]
- Ionic, covalent and co-ordinate covalent bond are present in:
(a) SO_2 (b) NH_4Cl (c) C_2H_2 (d) H_2O [FSD, GRW 11]
- The highest electronegative element in the periodic table is:
(a) Oxygen (b) Nitrogen (c) Chlorine (d) Fluorine [LHR 10]
- Bond angle between H-S-H bonds is:
(a) 105.5° (b) 107.5° (c) 92° (d) 95° [LHR 10]
- S.I Unit of dipole moment is:
(a) pm (b) Debye (c) mC (d) All [LHR 10]
- The amount of energy released by absorbing electron in the valence shell is:
(a) Ionization energy (b) Electron affinity (c) Electro-negativity (d) Atomization energy [MTN 08, 09, GRW 10]
- The number of electrons shared in SF_6 :
(a) 4 (b) 12 (c) 6 (d) 8 [GRW 10]

Bonding And Compoundscrossword Answers

**Charles M. Seiger, Edwin F.
Bartholomew, Frederic H. Martini**



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