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CHEMICAL REACTIONS

Chapter Test A

A. Matching

Match each description in Column B with the correct term in Column A. Write the letter of the correct description on the line.

Column A	Column B
1. coefficient	 a. a chemical equation that does not indicate relative amounts of reactants and products
2. spectator ion	b. a starting substance in a chemical reaction
3. combustion reaction	e. a list of metals in order of decreasing reactivity
4. reactant	 d. a reaction in which a single compound is broken down into simpler substances
5. skeleton equation	e. a whole number that appears before a formula in an equation
6. balanced equation	f. a new substance formed in a chemical reaction
7. activity series	g, a particle not directly involved in a chemical reaction
8. product	 an equation in which each side has the same number of atoms of each element
9. decomposition reaction	 i. a reaction in which the atoms of one element replace the atoms of a second element
10. single-replacement reaction	j. a reaction in which oxygen reacts with another substance, often producing heat or light

B. Multiple Choice

Choose the best answer and write its letter on the line.

- 11. The symbol Δ in a chemical equation means:
 - a, heat is supplied to the reaction.
 - b. a catalyst is needed.
 - c. yields.
 - d. precipitate.
- 12. A catalyst is:
 - a. a solid product of a reaction.
 - b. one of the reactants in a single-replacement reaction.
 - c. not used up in a reaction.
 - d. the product of a combustion reaction.

Chapter 11 1 Decscribing Chemical Reactions

F Rizvi

Chapter	11 1	Decscribing	Chemical	Reactions:

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