# 10

# CHEMICAL QUANTITIES

# Chapter Test A

# A. Matching

Match each description in Column B to the correct term in Column A. Write the letter of the correct description on the line.

12. The mass of a mole of NaCl is its:

b. empirical formula.

a. modar mass.

	Column A		Column B
1.	standard temperature and pressure		the volume occupied by a mole of any gas at STP (22.4 L)
2	molar volume	b.	the mass in amus of an atom of an element
3.	representative particle	( <b>C</b> )	the number of representative particles in a mole of a substance
4.	molar mass	di.	the percent by mass of each element in a compound
5.	atomic mass	·C.	the mass of a mole of a substance
6.	Avogadro's number	E.	the formula that gives the lowest whole-number ratio of the elements in a compound
7.	mole (mol)	85-	an atom, a formula unit, or a molecule
8.	empirical formula	b.	the mass in grams of a mole of any pure substance
9.	percent composition	30.	$6.02 \times 10^{23}$ representative particles of a substance
10.	molar mass	13-	0°C and 101.3 kPa
B. Multip	ole Choice		
Choose the be	est answer and write its letter	ore i	the time.
11.	How many hydrogen atoms C <sub>1</sub> H <sub>2</sub> O?	are	in 4 molecules of isopeopyl alcohol,
	a. $4 \times (6.02 \times 10^{23})$		c. 32
	b. 4		<b>d.</b> $32 \times (6.02 \times 10^{23})$

c. gram atomic mass.

d. atomic mass.

# **Chapter 10 Chemical Quantities Test Answer Key**

**American Medical Association** 

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**Soil Chemical Analysis** Marion LeRoy Jackson, 2005 General Chemistry Horace Grove Deming, 1923 Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction General, Organic, & Biological Chemistry Janice Gorzynski Smith, 2024-12 My goal in writing this text was to relate the fundamental concepts of general organic and biological chemistry to the world around us and in this way illustrate how chemistry explains many aspects of everyday life A key feature is the use of molecular art to illustrate and explain common phenomena we encounter every day Each topic is broken down into small chunks of information that are more manageable and easily learned Students are given enough detail to understand basic concepts such as how soap cleans away dirt and why trans fats are undesirable in the diet without being overwhelmed Metal-Containing Molecules and Nanomaterials Ana de Bettencourt-Dias, 2024-07-25 Since the discovery and successful deployment of salvarsan to treat syphilis as well as cisplatin as a cancer drug the field of metallodrugs has been flourishing In addition to therapeutic applications metals and metal containing molecules have properties such as an electrochemical or an optical signal that can be used to diagnose biologically relevant molecules or unravel cellular metabolism Moreover metals are part of the essential elements for life and of increased attention in nutraceuticals Most of the current diagnosis processes therapeutics and nutraceuticals are based on organic molecules MILS 26 reviews in 7 comprehensive chapters written by 20 experienced authors the most recent metal containing molecules and nanomaterials for diagnosis and therapy Features The field of metallodrugs has been flourishing and this area of research can be used to diagnose specific biologically relevant molecules or unravel cellular metabolism The authors are preeminent bioinorganic and medicinal inorganic chemists and review the most current research in this field The book discusses the properties and therapeutic applications of metals and metal containing molecules including the use of electrochemical or optical signal for diagnosis of biologically relevant molecules It endorses and stimulates research in the vibrant field of biological inorganic chemistry. It reviews the most current research supporting the relevance of metal containing molecules and nanomaterials for diagnosis and therapy Introductory Chemistry Charles H. Corwin, 2005 For one semester courses in Basic Chemistry Introduction to Chemistry and Preparatory Chemistry and the first term of Allied Health Chemistry This text is carefully crafted to help students learn chemical skills and concepts more effectively Corwin covers math and problem solving early in the text he builds student confidence and skills through innovative problem solving pedagogy and technology formulated to meet student needs Journal of the American Medical Association American Medical Association, 1925

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