## Chemistry

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#### (Chapter 10) (Heloakanes and Heloarenes) (Exercises)

#### (Class 12)

- (iii) CH<sub>3</sub>CH(Br)CH<sub>2</sub>CH<sub>3</sub> + NaOH  $\xrightarrow{water}$  CH<sub>3</sub>CH(OH)CH<sub>2</sub>CH<sub>3</sub> + NaBr 2-Bromobutane Butan-2-ol
- (iv) CH<sub>3</sub>CH<sub>3</sub>Br + KCN aq. Ethanole CH<sub>3</sub>CH<sub>2</sub>CN + KBr Bromobutane Nucleophilic Substitution Cyanoethane
- (vi) CH<sub>3</sub>CH<sub>2</sub>CH<sub>2</sub>OH + SOCl<sub>2</sub> → CH<sub>3</sub>CH<sub>2</sub>CH<sub>2</sub>Cl + SO<sub>2</sub> + HCl 1 - Propanol 1 - Chloropropane
- (vii) CH<sub>3</sub>CH<sub>2</sub>CH = CH<sub>2</sub> + HBr Peroxide CH<sub>3</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub> Br
  But -1-ene 1 Bromobutane
  Br
- (viii) CH<sub>3</sub>CH = C (CH<sub>3</sub>)<sub>2</sub> +HBr CH<sub>3</sub> CH<sub>3</sub> -CH<sub>2</sub> C CH<sub>3</sub>

  2-Methylbut-2-ene (Markovnikov addition)

#### 2-Bromo-2-methylbutane

#### Question 10.15:

Write the mechanism of the following reaction:

#### \_\_Answer 10.15:

The given reaction is:

The given reaction is an SN<sub>2</sub> reaction. In this reaction, CN<sup>-</sup>acts as the nucleophile and attacks the carbon atom to which Br is attached. CN<sup>-</sup> ion is an ambident nucleophile and can attack through both C and N. In this case, it attacks through the C-atom.

#### Question 10.16:

Arrange the compounds of each set in order of reactivity towards SN2 displacement:

- (i) 2-Bromo-2-methylbutane, 1-Bromopentane, 2-Bromopentane
- (iii) 1-Bromo-3-methylbutane, 2-Bromo-2-methylbutane, 2-Bromo-3- methylbutane
- (iii) 1-Bromobutane, 1-Bromo-2, 2-dimethylpropane, 1-Bromo-2-methylbutane, 1-Bromo-3-Methylbutane.

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Spencer L. Seager, Michael R. Slabaugh

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