

# Atomic Structure

Name: \_\_\_\_\_

Use the words in the box to complete the sentences.

Electron	Proton	Neutron	Atom
Quark	Compton	Radioactive	Isotope
Photon	Gamma	Alpha	Beta
Positron	Neutrino	Antineutrino	Antiparticle

1. An atom is made up of a central nucleus of protons and neutrons, with electrons orbiting the nucleus.

2. The nucleus is made up of protons and neutrons. Protons have a positive charge, while neutrons have no charge.

3. Electrons have a negative charge and are found outside the nucleus.

4. The mass of an atom is concentrated in the nucleus, which contains most of the atom's mass.

5. The size of an atom is determined by the distance between the nucleus and the outermost electrons.

6. The nucleus is made up of protons and neutrons, which are held together by the strong nuclear force.

7. Electrons are held in place by the electromagnetic force.

8. The overall charge of an atom is zero, as the number of protons equals the number of electrons.

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**Emilie Sanchez**



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