

Stoichiometry Worksheet #1

Mole-to-Mole Problems



How many moles of potassium nitrate are produced when 7.70 moles of aluminum nitrate react with excess potassium phosphate?

7.70 mol $\text{Al}(\text{NO}_3)_3$	$\frac{3 \text{ mol KNO}_3}{1 \text{ mol Al}(\text{NO}_3)_3}$	=	23.1 mol KNO_3
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How many moles of hydrogen are needed to completely react with 2.60 moles of nitrogen?

2.60 mol N_2	$\frac{3 \text{ mol H}_2}{1 \text{ mol N}_2}$	=	7.80 mol H_2
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How many moles of ammonia are produced with 11.30 moles of nitrogen?

11.30 mol N_2	$\frac{2 \text{ mol NH}_3}{1 \text{ mol N}_2}$	=	22.6 mol NH_3
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How many moles of oxygen are necessary to react completely with 4.0 moles of propane?

4.0 mol C_3H_8	$\frac{5 \text{ O}_2}{1 \text{ mol C}_3\text{H}_8}$	=	20 O_2
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Mole-to-Mass and Mass-to-Mole Problems



How many grams of barium chloride are needed to react completely with 5.00 moles of silver nitrate?

5.00 mol AgNO_3	$\frac{1 \text{ mol BaCl}_2}{2 \text{ mol AgNO}_3}$	$\frac{208.23 \text{ g BaCl}_2}{1 \text{ mol BaCl}_2}$	=	521 g BaCl_2
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Cloze Ing In On Science Worksheet Answers

Stoichiometry

Julia Burton



Cloze Ing In On Science Worksheet Answers Stoichiometry:

Advancing Scientific Literacy in Chemistry Julia Burton, 2011 The books in this series offer an attractive effective method for developing scientific literacy Contain engaging activities that will draw in even students who feel they cant do science Can be used to introduce or reinforce science concepts and vocabulary Cover a wide range of topics within each broad subject area of biology physics chemistry and earth science

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